Optoelectronic Characteristics of UV Photodetector Based on ZnO Nanopillar Thin Films Prepared by Sol-Gel Method

K. J. Chen¹, F. Y. Hung²,*, S. J. Chang¹ and S. J. Young¹

¹Institute of Microelectronics & Department of Electrical Engineering, Center for Micro/Nano Science and Technology, National Cheng Kung University, Tainan 701, Taiwan, R.O.China
²Institute of Nanotechnology and Microsystems Engineering, Center for Micro/Nano Science and Technology, National Cheng Kung University, Tainan 701, Taiwan, R.O.China

ZnO thin films were prepared on a quartz substrate by sol-gel method and a UV photodetector was constructed on the ZnO thin films, with a circular spiral structure in contact with 30 nm IrO₂ electrodes. The ZnO thin films were crystallized at various crystallized temperature (600℃~700℃) for 1 hour in pure oxygen atmosphere, and were then analyzed by X-ray diffraction (XRD) and the scanning electron microscopy (SEM) to investigate the thin film crystallized structures. From photoluminescence (PL) and I-V measurement, the 650℃ thin film not only possessed a better crystallization but also had nanopillar structures that revealed an excellent characteristic of UV photodetector.

(Received October 10, 2008; Accepted February 9, 2009; Published March 25, 2009)

Keywords: ZnO, thin film, nanopillar, iridium

1. Introduction

Photodetectors operating in the UV region are important devices that can be used in many commercial and military applications, such as ozone layer monitoring, flame detection and missile warning systems¹,² etc. In recent years, UV photodetectors have been fabricated on wide direct bandgap materials. Notably, ZnO possesses a wide direct band-gap energy of 3.37 eV and a larger exciton-binding energy of 60 meV,³ that have been used by several deposition techniques, such as RF magnetron sputtering, molecular beam epitaxy (MBE), metal organic chemical vapor deposition (MOCVD), pulsed laser deposition (PLD) and the sol-gel process.⁴,⁸ Among these methods, the sol-gel method enjoys the advantages of being able to prepare large area ZnO thin films at low cost and easy technology.⁹ Therefore, several investigations have prepared UV photodetectors using sol-gel synthesized ZnO thin films.¹⁰,¹¹ However, how the crystallization of sol-gel ZnO thin film affects the optoelectronic characteristics has still not been investigated. In addition, almost all the photodetector devices used interdigital (IDT) circular structures as contact electrodes,¹²,¹³ and the effect of the spiral configuration is worthy of further investigation.

To improve the quality of the photodetector, the metal electrodes need a higher work function and a higher thermal stability.¹⁴,¹⁵ But, many electrodes with high work function are not stable at higher temperatures. Iridium (Ir) metal has some advantages in terms of thermal (1000℃) and chemical stability.¹⁶,¹⁷ In addition, iridium oxide (IrO₂) also has some advantages, such as high work function (≥5 eV), excellent thermal stability and high transmittance,¹⁸,¹⁹ and so can be used in the electrodes of photodetectors.²⁰,²¹

For these reasons, this study used sol-gel derived ZnO thin films with an 30 nm IrO₂ electrode to fabricate the UV photodetector, on order not only to understand the effect of different crystallizations, but also investigate the contribution of the spiral electrode configuration.

2. Experimental Procedure

The ZnO thin films were deposited on the quartz substrate using the sol-gel method. To prepare the aqueous solution of ZnO, 2M zinc acetate dihydrate (Zn(CH₃COO)₂·2H₂O) was synthesized with a diethanolamine (DEA) in hydralan methanol (dry ≤ 0.01% water), and the molar ratio of DEA to zinc acetate was 1.²² The aqueous solution was stirred at 140℃ for 3 h to form a homogeneous and transparent solution. The spin coating method with a rotation rate of 3000 rpm was used to coat all substrates. After that, the film samples were dried at 200℃ for 10 min to evaporate the solvent and remove organic residuals, and then were naturally cooled to room temperature (10 times consecutively). Finally, the film samples were subjected to different crystallizing temperatures (600, 650 and 700℃) for 1 h under O₂ atmosphere with a flow rate of 50 sccm. The thickness of the ZnO thin film was approximately 250 nm which was measured by a dual-beam focused ion beam (FIB) (not given here).

The response area (contact area) of UV photodetector with the circular spiral electrode configuration was bigger than that of the interdigital electrode configuration.¹²,¹³ Therefore, UV photodetectors were fabricated and based on circular spiral metal-semiconductor-metal (MSM) structures. Iridium (Ir) film of thickness 30 nm was patterned onto the surface of sol-gel derived ZnO film by electron beam evaporation to serve as the metal contacts. The width and space of the contact electrodes were 300 μm, and the active area of the UV photodetector was 6000 × 4000 μm². Notably, the optical transmittance of IrO₂ was higher than that of Ir.¹⁸,¹⁹ In order to improve the quality of electrode, the Ir electrode contact was annealed at 500℃ for 10 min under an O₂ atmosphere with a flow rate of 50 sccm to form IrO₂. The schematic structure of the MSM UV photodetectors is shown in Fig. 1.

*Corresponding author, E-mail: fyhung@mail.mse.ncku.edu.tw

10.2320/matertrans.MER2008371

©2009 The Japan Institute of Metals

EXPRESS REGULAR ARTICLE
In addition, the crystalline structure was analyzed by thin-film X-ray diffraction (XRD) with CuKα radiation. The surface morphology of ZnO samples was obtained with a working distance of 5 mm, a beam current of 98 pA and a high speed voltage of 5.0 kV via attached the scanning electron microscopy (SEM) of a dual-beam focused ion beam (DB-FIB: Nova NanoLab 200). The photoluminescence (PL) measurement was used to analyze the optical properties of ZnO crystallization by 325 nm UV light from a He-Cd laser at room temperature. Also, the current-voltage (I-V) characteristics of the devices were measured by a HP 4145 under the dark and illumination. The top-illuminated spectral response was quantified using a 250 W xenon (Xe) arc lamp white light source through the monochrometer (The spot size of monochromatic light was about 3000 µm) and calibrated for a monochromatic detection in the range of 300–450 nm.

3. Results and Discussion

3.1 Structural characteristics

The XRD patterns of ZnO thin films with different crystallization temperatures (600, 650 and 700 °C) are shown in Fig. 2. All the ZnO thin films corresponded to the wurzite-type ZnO structure and preferred an orientation of (002). With increasing the crystallization temperature from 600 to 650 °C, the intensity of the (002) diffraction peak increased, revealing that the crystallization of the ZnO thin film was improved by an adequate supply of thermal energy. In general, the quality of the ZnO films was able to be improved by increasing the annealing temperature. However, the crystallization mechanism of the ZnO thin film began to deteriorate at 700 °C. The main reason is that variations in the grain boundary energy made some of the grains grow unusually and destroyed the crystallization of the ZnO thin film under higher crystallization temperatures.

For 650 °C ZnO film, the (002) diffraction peak at 2θ = 34.7° with a FWHM of 0.38° showed the excellent quality of the ZnO thin film. The surface morphology of the 650 °C ZnO thin film is shown in Fig. 3. The film shows a large amount of grain boundaries and a few micro-pore structures scattered on the surface. These micro-pore structures are certain to appear on the sol-gel synthesized ZnO thin films. Notably, some nucleus-like structure was found on the grain surface (Fig. 3). When the surface image was rotated from 0° to 52° using FIB (from top view to side view), the nucleus-like structure could be identified as ZnO nanopillars (Fig. 4). The length of the nanopillars was approximately ~90 nm and grew disorderly on the film matrix. This result was that the residual zinc reacted with oxygen to form the ZnO nuclei. Increasing the duration of crystallization, the ZnO nuclei individually grew to form the ZnO nanopillars. However, under higher crystallization temperatures, the unusual growth of the grains not only restrained the growth of the ZnO nanowires, but also destroyed the crystallization. This is why the 650 °C ZnO thin film possessed the best crystallization mechanism.

3.2 Optical and electrical properties

The 650 °C films had excellent crystallized structures with nanopillars. Photoluminescence (PL) was used to check its optical quality. Figure 5 shows the PL spectrum (room
temperature) of sol-gel derived ZnO thin film. In the spectrum, the ZnO thin film contains a strong UV emission band at 381 nm (3.25 eV) and a very weak green emission band at 510 nm (2.43 eV), which can be attributed to the recombination of free excitons and oxygen vacancies in the ZnO lattice. In addition, the full-width half-maximum (FWHM) of the UV emission band was 118 meV. These results indicate that the crystallization of the ZnO thin film was just as good as ZnO epitaxial film produced by radio frequency (RF) or plasma-assisted MBE.

The IrO$_2$/ZnO/IrO$_2$ MSM structure (a single-layer IrO$_2$ electrode) with a circular spiral configuration was used to evaluate the UV detector performance.

The I-V characteristic of the ZnO photodetector was measured in darkness and under photoillumination in Fig. 6. Under a 5 V applied bias, it was found that the dark current was $4.32 \times 10^{-9}$ A and the photocurrent was $5.11 \times 10^{-7}$ A. In other words, the photocurrent was about two orders of magnitude larger than the dark current. Similar results have been obtained in relevant reports. Notably, the photocurrent and dark current differed by less than one order of magnitude. It is clear that the electrical properties of the present thin film had been enhanced. In addition, Fig. 7 shows the spectral response of the photocurrent measurement on the MSM ZnO photodetector. For a light wavelength of 360 nm and 5 V applied bias, the response was 1.1 A/W and the cut-off occurred at 370 nm (Fig. 7). Although the cut-off was not sharp, the present structure had higher photocurrent value than two orders of magnitude from 370 to 420 nm.

The 600°C is lower that cannot help the growth of the nanopillars by sol-gel method. Notably, the 700°C film contained a very broad green emission which indicated that the electrons would be trapped by some defects (crystallized temperature is higher), resulted in the degeneration in the performance of photodetector. For reasons mentioned above, the 650°C film not only possessed nanopillars, but also had better crystallized structure that was more suitable for application in photodetector. In addition, it is noteworthy that the IrO$_2$ electrode with the circular spiral configuration had the originality and better electrical properties. Furthermore, we can be confident that the sensitivity of the sol-gel derived ZnO thin film with nanopillars in the UV region is good enough for applications as UV photodetector materials.
4. Conclusion

The 650°C sol-gel derived ZnO thin film not only possessed a better crystallized mechanism, but also had nanopillars on the film surface that were able to enhance the opto-electronic properties of the UV photodetector. For I-V measurement, the curve corresponded to the schottky metal-semiconductor contacts and the photo-generated current arrived at $5 \times 10^{-7}$ Ampere under a bias voltage of 5 V. In addition, the photocurrent was 2 orders of magnitude larger than the dark current, quite adequate for a UV photodetector.

Acknowledgements

The authors are grateful to National Cheng Kung University, the Center for Micro/Nano Science and Technology (NCKU Project of D97-2700) and NSC 97-2221-E-006-018/NSC 97-2622-E-006-009-CC3 for the financial support.

REFERENCES