Fracture Toughness Evaluation of Thin Fe–Al Intermetallic Compound Layer at Reactive Interface between Dissimilar Metals\textsuperscript{a,1}

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The fracture toughness of Fe–Al intermetallic compounds (IMCs), FeAl and Fe\textsubscript{2}Al\textsubscript{5}, that form as a thin layer on steel substrate was investigated. A model for evaluating the fracture toughness of a brittle thin layer on an elastoplastic substrate was applied, and the fracture toughness was evaluated from the thickness of the IMC layer and the crack interval in the IMC layer after uniaxial tensile testing. The phase and microstructure of the IMC layer were varied to investigate their effects on the fracture toughness of the IMC. The relationship between layer thickness and crack interval was in a good agreement with the theoretical model, and the fracture toughness was evaluated adequately using the model. It was clarified that FeAl has higher fracture toughness than Fe\textsubscript{2}Al\textsubscript{5}, and that fine-grained Fe\textsubscript{2}Al\textsubscript{5} has higher fracture toughness than coarse-grained Fe\textsubscript{2}Al\textsubscript{5}.

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1. Introduction

Recently, a number of methods for bonding dissimilar metals, including steel/Al bonding\textsuperscript{1,2} and steel/Mg bonding,\textsuperscript{3} have been studied in order to produce lightweight metallic composite materials. It has been found that a brittle Fe–Al intermetallic compound (IMC) layer is formed at the interface between such dissimilar metals. The formation of the same kind of Fe–Al IMC layer has also been reported in alumining and galvanizing processes. It is widely recognized that such a formation of the Fe–Al IMC layer at these interfaces influences the reliability of the composite materials by degrading the bonding strength of the interface between dissimilar materials and the adhesion of the coating layer. Therefore, it is very important to clarify the mechanical properties of these IMC layers. Oikawa et al. studied the relationship between the thickness of the Fe–Al IMC layer and the peel strength in steel/Al joints and reported that the peel strength decreases considerably when the thickness of the IMC layer exceeds 2 µm.\textsuperscript{1} Furthermore, Mukae et al. reported that the shear strength of the interface between steel and aluminum decreases markedly when the thickness of the Fe–Al IMC layer in aluminum-clad steel exceeds 1.5 µm.\textsuperscript{2}

Thus, it is generally believed that for practical purposes, the thickness of the Fe–Al IMC layer at the interface between dissimilar metals should be less than 1 µm. Even though it is clear that the thickness of the Fe–Al IMC layer influences the interface strength, the dominant factors that explain the relationship between the thickness of the Fe–Al IMC layer and the interface strength between dissimilar metals have not been clarified, partly because of the fact that the fracture toughness of the Fe–Al IMC layer formed at the interface is still unknown.

To measure the fracture toughness of a brittle material such as an IMC, the indentation fracture method has been widely applied. In this method, the fracture toughness is evaluated from the crack length induced by forcing an indenter tip into the surface of a brittle material.\textsuperscript{4} Recently, the microscale evaluation of fracture toughness has been attempted using a nanoindenter.\textsuperscript{5} However, the crack size is still in the order of microns to allow for an accurate measurement even with a nanoindenter, which means that it is difficult to evaluate the fracture toughness of brittle thin materials with thickness in the order of submicrons.

In this study, we propose a method for evaluating the fracture toughness of thin Fe–Al IMC layers formed at the interface between dissimilar metals, applying fracture mechanics. The method enables us to evaluate the fracture toughness of brittle materials on a submicron scale, and consequently, to analyze the relationship between the fracture toughness and both the composition and the microstructure of Fe–Al IMC layers.

2. Theory

In a brittle thin layer formed on a substrate, a tensile stress is induced as a result of either substrate deformation or residual stress, such as lattice-misfit distortion and thermal stress. As a result of the tensile stress, various configurations of cracks form.\textsuperscript{6–13} It is known that when the substrate is subject to a uniaxial tensile deformation, steady-state cracks begin to propagate in the brittle layer in the direction perpendicular to the maximum principal stress as the applied stress reaches the critical stress \(\sigma_c\). The density of these steady-state cracks increases further with increasing tensile strain, and ultimately the crack interval asymptotically approaches a constant value \(\lambda_c\).\textsuperscript{7–13} In a metal substrate, plastic deformation occurs just below the crack, and the plastic zone is expected to be small if the substrate is sufficiently hard. In this case, it has been reported that the critical stress \(\sigma_c\) and the final crack interval \(\lambda_c\) can be given as...
a function of the film thickness $h$ on the basis of the linear elastic fracture mechanics model.\(^8,12\) On the other hand, if the substrate is soft, the plastic zone induced in the substrate expands and eventually covers the entire substrate surface. In this case, it has been quantitatively shown that the critical stress $\sigma_c$ and the final crack interval $\lambda_c$ can be given as functions of the brittle layer thickness $h$ using the shear lag limit model, in which it is assumed that the substrate is perfect elastoplastic body.\(^7,13\) In this study, the latter model is more appropriate, because our metallic substrates are very ductile.

In the shear lag limit model, it is assumed that the substrate material is a perfect elastoplastic body. In this case, as shown in Fig. 1, plastic deformation occurs in the substrate while steady-state cracks extend in the direction perpendicular to that of the maximum principle stress in the brittle layer. The plastic zone is concentrated near the interface with the brittle layer, and large slip deformation accumulates only in the plastic zone. Then, the plastic deformation releases part of the stress $\sigma$ acting on the brittle layer, and the stress component perpendicular to the crack surface becomes zero on the crack surface. In addition, the shear stress $\tau_Y$ at the interface between dissimilar metals takes a constant value, $\tau_Y = \sigma_Y / \sqrt{3}$, over the whole interface between the dissimilar metals, where $\sigma_Y$ is the yield strength of the substrate.

Therefore, the energy release rate $\Delta G$ upon crack extension can be given by subtracting the energy dissipated by interface sliding $\Delta W_d$ from the sum of the work done by the applied load $\Delta W$ and the change in strain energy released by interface sliding $\Delta U_s$.\(^7,13\)

$$\Delta G = \Delta W + \Delta U_s - \Delta W_d = \frac{\sigma_c h^2}{3E\tau_Y} = h\Gamma$$  \hspace{1cm} (1)

Here, $E$ is Young’s modulus of the brittle layer and $\Gamma$ is the fracture toughness of the brittle layer. On the other hand, the force balance at the interface is given by

$$\sigma h = \lambda \tau_Y / 2$$  \hspace{1cm} (2)

where $\lambda/2$ is the length of the plastic deformation region in the substrate. Therefore, the stress $\sigma_c$, that can induce steady-state cracks under the general yielding conditions is given by

$$\sigma_c = \sqrt{\frac{3E\Gamma\tau_Y}{h}}$$  \hspace{1cm} (3)

Finally, the saturated crack interval $\lambda_c$ is given by the minimum length of the plastic deformation region that can induce the tensile stress $\sigma_c$ in the brittle layer.

$$\lambda_c = 2 \left( \frac{9E\Gamma h^2}{\sigma_c^2} \right)^{1/3}$$  \hspace{1cm} (4)

It is noted that in the above discussion, we have assumed that the substrate is a perfect elastoplastic solid. However, the actual crack interval $\lambda_c$ is expected to deviate from eq. (4) because of the influence of work hardening of the substrate and the difference between the elastic constants of the substrate and those of the thin layer. The influences of work hardening and the different elastic constants on the energy release rate $\Delta G$ were numerically verified by Beuth and Klingbeil.\(^13\) It was shown that when the elastic constant of the substrate and that of the thin layer satisfy $\alpha = E_c/E_s < 0.5$, it is possible to estimate $\Delta G$ using the shear lag limit model with the average yield stress $\sigma_{Yeff}$ at the interface (effective yield stress) as a substitute for the yield strength $\sigma_Y$ of the substrate. In addition, the results of the numerical computation show that the effective yield stress $\sigma_{Yeff}$ does not depend on the length of the plastic deformation region and is almost constant as a function of the work hardening exponent.

In this study many steady-state cracks are generated in the IMC layer by loading various magnitudes of uniaxial tensile load on the steel substrate. Fe–Al IMC layers with various compositions and microstructures are formed, and the corresponding crack intervals $\lambda_c$ are measured. At the same time, the yield stress $\sigma_{Yeff}$ near the interface is measured with a nanoindenter, and finally, the fracture toughness is examined with respect to the composition and microstructure using eq. (4).

3. Experimental

To form uniform IMC layers with different compositions and microstructures, three different methods, namely, the reactive transient liquid phase (TLP) bonding method, the hot-dipping method, and the infiltration method, were used. The chemical compositions of the steel substrates and Mg alloys used in this study are shown in Tables 1 and 2, respectively.

In the reactive TLP bonding method, pure Ag with a thickness of 1 µm was first deposited onto the surfaces of interstitial-free (IF) steel and SUS304 by the electron-beam deposition method. The steel and Mg alloy were stacked so that the steel/Ag/Mg/Ag/steel layer structure was achieved, and then the stacked specimen was heated at 773 K for 3000 to 30000 s with a compression stress of 1 MPa in argon gas atmosphere, followed by furnace-cooled.

In the hot-dipping method, the 84 mol% Mg–16 mol% Ag eutectic alloy melt was prepared with an additional 3 to 4% Al. In a crucible, the IF steel was immersed in the Mg–Ag eutectic alloy melt at 773 K for 2000–5000 s in argon gas atmosphere and then water-cooled.

In the infiltration method, different Mg–Al alloys with Al concentrations of 1, 2, 3 and 4 mol% were prepared by mixing pure Mg and Mg alloy AZ63. Stacked IF steel sheets with a space and Mg–Al alloy were heated at 973 K in a
crucible. The molten Mg-Al alloy infiltrated between the stacked IF steel sheets upon applying argon gas pressure. Heating lasted for 900 s, and was followed by then air-cooling.

Note that the layered structures of the samples have a triple-layered structure of steel/Mg alloy/steel in the reactive TLP bonding and infiltration methods and a double-layered structure of steel/Mg–Ag–Al eutectic alloy in the hot-dipping method. A sample of 30 mm length × 6 mm width was cut from each sample for uniaxial tensile tests. The interfacial IMC layers of the samples after a tensile test were observed from the direction perpendicular to the tensile direction using a field emission scanning electron microscope (FE-SEM). The crack interval \( \lambda \) was obtained as the average of more than 50 cracks. At the same time, we identified the IMC phase using X-ray diffraction (XRD), performed composition analysis by energy-dispersive X-ray spectroscopy (EDS), and analyzed the IMC microstructure using electron backscatter diffraction (EBSD). A nanoindenter was used to measure the hardness of the steel substrate near the IMC.

4. Results

4.1 Formation of Fe–Al IMCs

Cross-sectional SEM images of the Fe–Al IMC obtained by each method are shown in Fig. 2. It was confirmed that a thin and uniform IMC film layer was formed by all the fabrication methods. Figure 3 shows X-ray diffraction spectra obtained from the IMC layers. It was confirmed that Fe2Al5 mainly formed in the reactive TLP bonding method and the hot-dipping method, whereas Fe2Al5 and FeAl formed in the infiltration method. Figure 4 shows an enlarged image of Fig. 2(c). The EDS results show that the formation of FeAl dominates the IMC layer by the infiltration method and that Fe2Al5 only partly covers the surface of the IMC layer. Therefore, we regard the samples made by the infiltration method as having the single phase of FeAl and disregard Fe2Al5.

4.2 Crack interval

Figure 5(a) shows a cross-sectional image of the interface of the sample at a tensile strain of 10%, and Fig. 5(b) shows a plan-view image of the IMC layer after the Mg alloy was

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Table 1 Chemical compositions of steel substrates (mass)%.

<table>
<thead>
<tr>
<th>Materials</th>
<th>C</th>
<th>Si</th>
<th>Mn</th>
<th>P</th>
<th>S</th>
<th>Ni</th>
<th>Cr</th>
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<tr>
<td>SUS304</td>
<td>0.04</td>
<td>0.66</td>
<td>1.05</td>
<td>0.032</td>
<td>0.002</td>
<td>8.57</td>
<td>18.16</td>
</tr>
<tr>
<td>IF Steel</td>
<td>0.001</td>
<td>0.16</td>
<td>0.22</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Table 2 Chemical compositions of magnesium alloys (mass)%.

<table>
<thead>
<tr>
<th>Materials</th>
<th>Al</th>
<th>Zn</th>
<th>Mn</th>
<th>Si</th>
<th>Cu</th>
<th>Ni</th>
</tr>
</thead>
<tbody>
<tr>
<td>AZ31</td>
<td>2.5–3.5</td>
<td>0.6–1.4</td>
<td>0.2–1.0</td>
<td>&lt;0.005</td>
<td>&lt;0.1</td>
<td>&lt;0.05</td>
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<tr>
<td>AZ61</td>
<td>5.8–7.2</td>
<td>0.04–1.5</td>
<td>0.15–0.5</td>
<td>&lt;0.005</td>
<td>&lt;0.1</td>
<td>&lt;0.05</td>
</tr>
<tr>
<td>AZ63</td>
<td>5.3–6.7</td>
<td>2.5–3.5</td>
<td>0.15–0.35</td>
<td>0.3</td>
<td>0.25</td>
<td>0.01</td>
</tr>
</tbody>
</table>

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Fig. 2 Cross-sectional images of IMC made by (a) reactive-TLP bonding, (b) hot-dipping method and (c) infiltration method.

Fig. 3 XRD patterns from the steel surfaces after (a) reactive-TLP bonding, (b) hot-dipping and (c) infiltration.

Fig. 4 Close-up image of IMCs shown in Fig. 2(c).
dissolved in HNO₃ for the same sample. Figure 5 confirms that cracks extend linearly in the direction perpendicular to the tensile direction with regular intervals.

The dependence of the crack interval on tensile strain was also evaluated. Figure 6 shows the relationship between crack interval and tensile strain for the samples (IMC thickness 1.4 ± 0.25 µm) prepared by the reactive TLP bonding method. It is observed that crack interval remains almost constant among the samples subjected to strains of more than 4%. This suggests that when the applied strain is more than 4%, yielding of the steel substrate occurs entirely near the interface, and then no more cracks form. Therefore, to study the relationship between the thickness of the IMC layer and the crack interval, a tensile strain of 10% is applied to all samples in this study so that the crack interval is fully saturated.

It should be noted that the Mg alloy layer was not removed from the tensile specimens fabricated by the reactive TLP method and infiltration method, as shown in Fig. 5, even though only the steel substrate and IMC layer are assumed in the shear lag limit model, as shown in Fig. 1. The reason for this is that since the fracture toughness Γ is proportional to the square of the yield strength of the substrate, as eq. (4) shows, the effect of the Mg alloy, which has lower yield strength, on the crack interval is small. As discussed in the following section, work hardening is not observed in the Mg alloy in the vicinity of the IMC layer, although it is pronounced in the steel near the interface. Therefore, it is considered that the Mg alloy does not contribute to the stress accumulation in the IMC layer. Thus, we ignore the influence of the Mg alloy. In fact, crack interval was quite similar between specimens with Mg alloy and those from which Mg alloy was removed by HNO₃.

4.3 Deformation of steel substrates

To investigate the work hardening behavior of steel substrates near the interface, the plastic deformation region was analyzed using EBSD and hardness measurement with a nanoindenter for SUS304 and IF steel at a tensile strain of 10%. The results are shown in Fig. 7(a) for SUS304 and Fig. 7(b) for IF steel. The left figure shows the distribution of the kernel average misorientation (KAM) values of the steel substrates, and the right figure shows the hardness distribution over the corresponding depth. Hardness in the figure

![Fig. 5 Cracks observed after uniaxial tensile testing with 10% strain (a) cross-sectional view and (b) plan view.](image)

![Fig. 6 (a) Relationship between nominal strain and crack interval. Cross-sectional images of cracks in IMC after applying (b) 2%, (c) 4% and (d) 10% tensile strain.](image)
was normalized by the hardness of the base steels distant from the interface. The vertical axis shows the distance of the measurement point from the interface, which corresponds to the left figures. The increase in the KAM values near the interface, as shown in Fig. 7(a), confirms that the steel substrate exhibits a large plastic deformation near the interface where the hardness is correspondingly high. In addition, the results of the hardness test include those measured in the samples with different IMC thicknesses. It is clear that the hardness distribution is almost the same regardless of IMC layer thickness, which supports the result of the numerical analysis by Beuth and Klingbeil. A similar tendency was seen for the IF steel in Fig. 7(b), but compared with the results of for SUS304, hardness rises more steeply in the region closer to the IMC layer. This confirms that the region of plastic deformation is closer to the interface than it is in SUS304 and that IF steel deforms more significantly than SUS304 near the interface.

### 4.4 Relationship between IMC thickness and crack interval

Figure 8 shows plots of the relationship between IMC thickness and crack interval for the samples with tensile strain of 10%. It was confirmed that the crack interval increases with IMC thickness, and even at the same thickness level, there exist several samples whose crack intervals deviate from others, such as those marked with (b) and (c). To investigate the influence of the microstructure, the microstructure of the IMC was analyzed using EBSD. Figure 9 shows the IQ-map obtained from the samples corresponding to samples (a)-(d) in Fig. 8. It is clear that the microstructure in Fig. 9(c) is different from those of other samples.
Figure 10 shows the grain size distribution of the samples shown in Fig. 9. Figures 10(a), 10(b) and 10(d) show similar grain size distributions, where the average grain size is approximately the same. Although the fabrication methods used to produce the samples in Figs. 10(b) and 10(d) were different, the grain size distribution is approximately the same as that shown in Fig. 8. Therefore, we can still regard them as having identical microstructures. On the other hand, in the sample in Fig. 10(c), which contains grains longer than 1 µm, the average grain size was greater than those in the other types of samples. In the samples composed primarily of Fe2Al5, we either find Fe2Al5 with a uniform grain size of 500 nm or less, as shown in Figs. 10(a), 10(b) and 10(d), or Fe2Al5 with many coarse grains of sizes greater than 1 µm, as shown in Fig. 10(c). They are called fine- and coarse-grained Fe2Al5, respectively, in the following discussion. Figure 11 shows a logarithmic plot of the relationship between the thickness and the crack interval for FeAl and the fine- and coarse-grained Fe2Al5. Since the crack interval \( \lambda_c \) is proportional to the \( 2/3 \) power of the IMC layer thickness \( h \) according to eq. (4), it is expected that, in the logarithmic representation, the relationship between the thickness and crack interval will be a straight line with a slope of \( 2/3 \). In our experiments, it was confirmed that a linear relationship between the thickness and crack interval holds for all microstructures and compositions, with a slope of approximately \( 2/3 \).

4.5 Evaluation of fracture toughness

Fracture toughness can be evaluated from the intercept of the straight lines in Fig. 11 for each respective microstructure using eq. (4). Young’s modulus of 260 GPa for FeAl\(^{14}\) and of 220 GPa for Fe2Al5\(^{15}\) were used. In addition, the stress estimated from the results of the hardness test was used as the effective yield stress \( \sigma_{yeff} \) of the substrate. As a result, fracture toughness was estimated to be 1.1 MPa \( \cdot \) m\(^{1/2} \) for FeAl, 0.51 MPa \( \cdot \) m\(^{1/2} \) for fine-grained Fe2Al5 and 0.26 MPa \( \cdot \) m\(^{1/2} \) for coarse-grained Fe2Al5.

Previously reported measurements of bulk materials have revealed that the fracture toughness is 5 MPa \( \cdot \) m\(^{1/2} \)\(^{16}\) for Fe–Al (40% Al) and 0.34 MPa \( \cdot \) m\(^{1/2} \)\(^{17}\) for Fe2Al5. It is difficult to directly compare these values, because chemical compositions and microstructures of the IMCs used in the previous measurements differ from those in our study. However, the fracture toughness values obtained are roughly the same as those reported previously, and it can be said that the fracture toughness seems to have a similar dependence on the phase.

5. Conclusions

In this study, the fracture toughness of the brittle thin Fe–Al IMC layer formed at the interface between dissimilar metals was estimated by a method based on fracture mechanics, considering the plastic deformation of the substrate. The following conclusions were drawn in this study.
The relationship $\lambda_c \propto h^{2/3}$ holds between the thickness $h$ of the brittle thin layer and the crack interval $\lambda_c$, and the fracture toughness was evaluated by measuring $\lambda_c$.

The experimentally obtained fracture toughness values were 1.1 MPa·m$^{1/2}$ for FeAl, 0.51 MPa·m$^{1/2}$ for fine-grained Fe$_2$Al$_5$ and 0.26 MPa·m$^{1/2}$ for coarse-grained Fe$_2$Al$_5$. These results indicate that higher fracture toughness can be obtained by using an iron-rich phase and finer grained microstructures.

REFERENCES